

# CLEANING LABORATORY EVALUATION SUMMARY

SCL #: 2000

DateRun: 02/08/2000

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ClientType: Chemical Light mfr

ProjectNumber: Project #1

Substrates: Liquid

PartType: Part

Contaminants: Phthalates

Cleaning Methods:

Analytical Methods: Light Meter

Purpose: To identify a methodology for determining contamination levels of cleaning solution and rinse waters.

Experimental Procedure: SUBSTRATE MATERIAL: Liquid-(water)  
CONTAMINANTS: Activator compound (75-65-0, 131-11-3, 7722-84-1); Phthalate (84-74-2)

Results: Two series of fluorescing standards were made using the two components of the chemical lights. One set of standards were diluted up to 100 ml with DI water in beakers. The second set was added to 100 ml in beakers. Each standard was stirred with a glass rod prior to recording light intensity. Table 1 lists the standards used for evaluation.

Table 1. Standard Chemical Light Mixtures

| SET 1 | %Contaminant | 1 | 2.5 | 5 | 10 | 25 | 50 | 100 |
|-------|--------------|---|-----|---|----|----|----|-----|
| Set 2 | # Drops      | 1 | 2   | 3 | 4  | 5  | X  | X   |

For Set 1, a DI blank was placed into a black light chamber. The SPER Light Meter probe was placed into the chamber and the light intensity was recorded in foot candles. Set 1 standards were placed one at a time into the chamber in the exact place as the blank. Light intensity readings were made from the same place every time. Readings were then measured in a dark room with no black light chamber. Analysis was performed at two sites on the beaker for Set 1, the side and the top, and only one site for Set 2, the top.

A final visual observation was made using the black light chamber. Set 1 standards were all placed into the chamber and the coloring of the mix was observed. The four client supplied wash/rinse water samples were also analyzed in this manner to determine relative levels of contamination.

After recording all readings of the two sets of standards, correlation factors were determined using Microsoft Excel LINEST function. From this data, graphs were made to illustrate light meter readings versus the amount of contaminant in the standard. The data from Set 1 was best represented by the readings taken from inside the black light chamber. The correlation was found to be 0.9740 (1.0 being ideal). When the Natural Log of the values were taken, the readings from side of the beaker outside the chamber yielded the highest correlation, 0.9921. Table 2 lists the Light Meter readings and the corresponding correlations. Figure 1 shows the natural log of the data along with the Best Fit line based on the LINEST calculations.

Table 2. SET 1 Readings

|                     |         |       |       |
|---------------------|---------|-------|-------|
| Inside Chamber      |         |       |       |
| Outside             | Chamber |       |       |
| Contaminant Reading | Top     |       |       |
| Side                | Reading |       |       |
| % by vol            | Inside  | Side  | Top   |
| 0                   | 0.12    | 2.13  | 0.51  |
| 1                   | 0.21    | 1.75  | 0.92  |
| 2.5                 | 0.23    | 2.03  | 0.99  |
| 5                   | 0.32    | 1.92  | 1.14  |
| 10                  | 0.25    | 2.31  | 1.42  |
| 25                  | 0.36    | 3.06  | 2.40  |
| 50                  | 0.51    | 4.30  | 4.00  |
| 100                 | 1.20    | 13.58 | 14.83 |
| Correlation         | 0.97    | 0.96  | 0.96  |
| LN Correlation      | 0.932   | 0.99  | 0.97  |

For Data Set 2, the readings taken from the top of the beakers outside the chamber yielded the highest correlation, 0.9971. Table 3 and Figure 2 show the data and correlations for Data Set 2.

Table 3. Set 2 Data

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| Contaminant | Inside   | Outside Top |
|-------------|----------|-------------|
| # Drops*    | Reading  | Reading     |
| 0           | 0.19     | 0.07        |
| 1           | 0.24     | 0.1         |
| 2           | 0.26     | 0.12        |
| 3           | 0.23     | 0.14        |
| 4           | 0.31     | 0.16        |
| 5           | 0.49     | 0.18        |
| Correlation | 0.839    | 0.997       |
| * 1 drop    | 0.047 ml |             |

Visual observations of both sets of data revealed that the higher the contaminant concentration was the more yellow the solution was. Table 4 lists the observations made for both sets of standards. Of the four client supplied samples, two were identified to contain low levels of contamination. However, the amount of fluorescing was far less than the 1 drop sample from the Second Set. This may relate to a low volume of contaminant in the solution, but before concluding such, a fresh sample should be evaluated to eliminate the time variable (sample has been sitting for over a month).

Table 4. Visual Inspection of Fluorescence

| SAMPLE       | OBSERVATION   |
|--------------|---|
| Set 1        |   |
| 1            | Green glow  |
| 2.5          | Green glow, yellow globs  |
| 5            | Green, yellow glow-yellow ring on bottom                              |
| 10           | Green, yellow glow- thin yellow ring on bottom and top                |
| 25           | Green, yellow glow-layers of each, more green than yellow             |
| 50           | Yellow glow, small green ring in middle                               |
| 100          | Yellow glow, small thin green ring on top                             |
| Set          |   |
| 2, 1-5 drops | Green glow, slight increase in intensity as number of drops increases |
| Basket       | Green glow, no yellow, particulate matter floating                    |
| Water Wash   | Green glow-faint  |
| Water Rinse  | No color  |
| DI Rinse     | No color  |

Summary:

Conclusion:

The use of black light fluorescence was found to be a possible way for method for determining the contamination levels of rinse/wash water. Using a light meter may aid in determining the quantitative levels of contaminants in these solutions.